¹³C-NMR spectra of chlorophylls a and b in green algae biosynthetically enriched with [1-¹³C]glutamate

Tamiko Oh-Hama, Haruo Seto, Noboru Otake and Shigetoh Miyachi

Institute of Applied Microbiology, University of Tokyo, Bunkyo-ku, Tokyo 113, Japan

Received 20 December 1982; revision received 27 January 1983

Abstract not received

5-Aminolevulinic acid Chlorophyll a Chlorophyll b [1-13C]glutamic acid Scenedesmus obliquus

1. INTRODUCTION

5-Aminolevulinic acid (ALA), a precursor of the tetrapyrrole compounds, can be formed either by condensation of succinate and glycine which is catalyzed by ALA synthetase, or from the entire carbon skeleton of glutamate (C-5 pathway [1]). In the former, C-5 of ALA is derived from C-2 of glycine, while it is derived from C-1 of glutamate in the C-5 pathway. We reported that ¹³C of [1-¹³C]glutamate was preferentially incorporated into the carbon atoms of chlorophyll derived from C-5 of ALA, while that of [2-¹³C]glycine was into methoxyl group adjacent to the alicyclic ring, in green alga, Scenedesmus obliquus in light [2]. However incorporation pattern of ¹³C in chlorophylls a and b has not been clarified.

Here, 13 C-NMR spectra of chlorophylls a and b enriched with $[^{13}$ C]glutamate indicated that the incorporation of carbon into four methine bridge carbon atoms (C_{α} , C_{β} , C_{γ} and C_{δ} , corresponding to C-5, C-10, C-15 and C-20, respectively, according to IUPAC-nomenclature) and four sp_2 quaternary carbon atoms (C-12, C-14, C-16, C-17), derived from C-5 of ALA, was much higher in chlorophyll a (fig. 1) than in b. Since increase of chlorophyll b was almost equivalent to that of a, we assumed that chlorophyll b was not only syn-

Abbreviation: ALA, 5-aminolevulinic acid

the sized from newly formed chlorophyll a, but also from that had been produced before feeding labeled glutamate.

2. MATERIALS AND METHODS

2.1. Labeling of chlorophylls a and b

Scenedesmus obliquus cells were cultured in the dark for 4 days at 29°C [4]. The algal cells were then suspended in 400 ml phosphate buffer (4 mM,

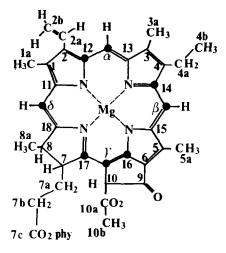


Fig. 1. The structure and numbering system of chlorophyll a. Atoms denoted by closed circles (•) are derived from the C-5 of ALA [3].

pH 6.5) containing 8 mM KNO₃ and 1 mM MgSO₄ at 5 ml packed cell vol./l. After 2 h-illumination with daylight fluorescent lamps (10 kLux), $[1^{-13}C]$ glutamate (90 atom%) was added to give 1 mM final conc. The algal cells were illuminated (Xe-lamp, 12 kLux) for 6 h [2], then collected to isolate chlorophylls a and b. Mixture of chlorophylls a and b was also isolated after feeding $[1^{-13}C]$ (90 atom%)- and $[1^{-14}C]$ glutamate for 5 h [2].

Chlorophylls a and b were determined spectrophotometrically [5]. Chlorophyll was isolated as in [2], using DEAE-Sepharose 6B [6]. Chlorophyll a was then separated from b using Sepharose 6B [6]. Radioactivity in chlorophyll fraction was counted with a windowless gas-flow counter.

2.2. ¹³C-NMR spectra

The spectra of chlorophyll in acetone- d_6 were recorded using a JEOL-FX 400 spectrometer under the following conditions: resonance frequency 100 MHz, pulse width 8 μ s, repetition time 2.34 s, spectral width 24 000 Hz, 20 500-24 000 transients. Chemical shifts are expressed in ppm relative to TMS. They were derived from the acetone- d_6 methyl-signals as internal standard (29.2 ppm from TMS). Amounts of chlorophyll used for analysis were indicated in each legend.

[1-13C]Glutamate was prepared as in [2]. [1-14C]-Glutamate was from New England Nuclear (Boston MA). DEAE-Sepharose CL-6B and Sepharose 6B were from Pharmacia (Uppsala). Acetone-d₆ was a product of Merck (Darmstadt).

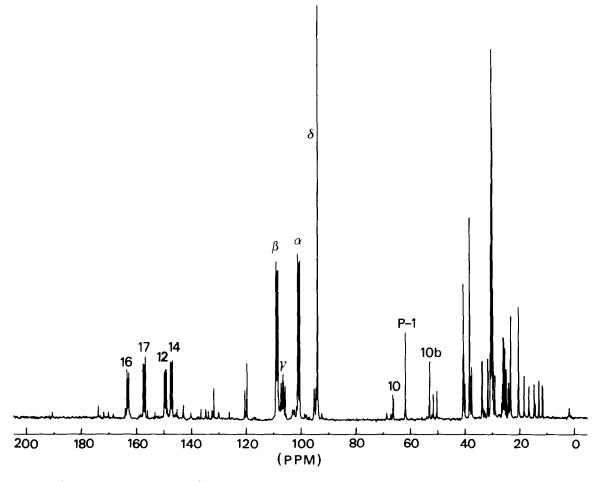


Fig. 2. The 13 C-NMR spectrum of chlorophyll a in acetone d_6 . 13 C-Enriched chlorophyll a (8.7 mg) was obtained by feeding [1- 13 C]glutamate to the cells of *Scenedesmus obliquus* for 6 h.

3. RESULTS AND DISCUSSION

When the dark-grown cells of *Scenedesmus* were illuminated with white light (12 kLux), chlorophyll content increased, doubling within several hours [4]. Chlorophylls a and b were isolated from the algal cells which had been fed [1- 13 C]glutamate for 6 h, following the pre-illumination which lasted for 2 h. Fig. 2 represents 13 C-NMR spectrum of 13 C-enriched chlorophyll a. The spectrum is similar to that of the total chlorophyll fraction enriched with 13 C under similar conditions [2], except that two rather minor peaks in sp_6 methine region (108 and 112 ppm) were eliminated. In [7] assignment

of two quaternary carbons, C-16 and C-17, was revised as shown in fig. 2, using a technique of long-range selective ¹H decoupling [7]. The present results indicate that ¹³C of glutamate was specifically incorporated in eight carbon atoms in chlorophyll a derived from C-5 of ALA. In [8], ¹³C of [1-¹³C] glutamate was incorporated into the chlorophyll macrocycle in a manner consistent with the operation of the C-5 pathway in excised, etiolated maize leaves.

Fig. 3 represents the 4-fold expanded spectra of chlorophylls a and b in the protonated and quarternary sp_2 carbon regions. Signals from most of the ¹³C-enriched carbon atoms of both chloro-

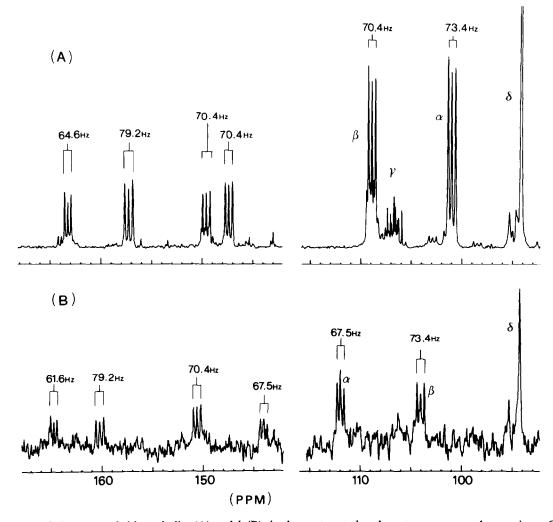


Fig. 3. Expanded spectra of chlorophyll a (A) and b (B), in the protonated and quaternary sp_2 carbon regions. Chlorophyll a (8.7 mg) and chlorophyll b (left, 2.4 mg; right, 1.3 mg).

phylls a and b were split into doublets. The splitting patterns and coupling constants of these carbons are shown in fig. 3. Chemical shifts at 93.0, 99.9, 105.6 and 107.8 ppm for chlorophyll a correspond to δ , α , γ and β carbon atoms, and those at 93.1, 102.8 and 110.7 ppm for chlorophyll b to δ , β and α carbon atoms, respectively, according to early assignments of 'H-NMR of methylpheophorbide b and pheophytin b [9] and 13 C-NMR of chlorophyll b [10]. Doublets of 4 quarternary carbon atoms enriched with 13 C in chlorophyll a are centered at 146.3 (C-14), 148.5 (C-12), 156.1 (C-17) and 162.1 ppm (C-16); and those of chlorophyll b are of 142.9, 149.5, 159.1 and 163.7 ppm, respectively. In both pigments the ratios between the intensities of the doublet and singlet are the same (i.e., $\sim 2:1$) indicating that the relative incorporation of 13 C into sp_2 carbons in chlorophylls a and b molecules are the same. Multiple splitting was observed with the γ -methine carbon in chlorophyll a, which can arise from interactions with ¹³C-16 and 13 C-17. The multiple splitting in chlorophyll b would have caused vague multiplet patterns of the signals as shown in fig. 3B.

Chlorophyll b is synthesized from chlorophyll a [11]. With Scenedesmus cells incorporation of labeled carbon atoms into chlorophylls a and b were studied in the presence of [1-14C]- and [1-13C]glutamate. Radioactivity in the chlorophyll fraction increased linearly for 5 h from the beginning, and chlorophyll b was produced during the second half period of the labeling time. Fig. 4 represents ¹³C-NMR spectrum of chlorophyll thus labeled after 5 h. It is clear that ¹³C of glutamate was 5-6-times greater in C_{α} and C_{β} of chlorophyll a than those of b. Chlorophyll b, however, comprised 40% of the total chlorophyll produced (table 1). Specific radioactivity in chlorophyll a produced during incubation with [1-14C]glutamate was 4-times greater than that in b, indicating higher dilution in b than in a. This indicates that chlorophyll b was not only synthesized from newly formed chlorophyll a, but also from that having been produced before feeding [13C]glutamate.

Chlorophyll b would be finally incorporated into light-harvesting chlorophyll a/b protein [12]. Shlyk [11] has proposed that biosynthesis of chlorophyll b can be synthesized only from a pool of newly synthesized chlorophyll a molecules. Consistently with that proposal, it was reported [13] that,



Fig. 4. ¹³C-NMR spectrum of chlorophyll enriched with ¹³C by feeding [1-¹³C]glutamate to algal cells for 5 h: α , β , γ and δ are methine bridge carbons of chlorophyll a; α' and β' , those of chlorophyll b; 8.4 mg chlorophyll was used.

when leaves of *Phaseolus* accumulating [14C]-chlorophyll a by intermittent illumination were transferred to continuous light in the absence of [14C]ALA, the leaves produced chlorophyll b, in which no 14C existed. In contrast, chlorophyll a, pre-accumulated in cucumber seedlings by intermittent illumination, was continuously converted to chlorophyll b during the subsequent dark periods provided CaCl₂ was present [14]. It is known that chlorophyll a formation in green algae proceeds in total darkness, while that in angio-

Table 1

Contents of chlorophylls a and b in dark-grown cells of *Scenedesmus* and increments of those after exposure of light in the absence, then in the presence of $[1^{-13}C]$ - and $[1^{-14}C]$ glutamate

	Chlorophyll (µmol)	
	a	b
Dark-grown cells		
(µmol/ml cells)	3.67	1.55
Produced during 2 h		
pre-illumination	0.9	n.d.
Produced during 5 h		
labeling	1.87	1.39

n.d., not detectable

sperms requires light. Results described above suggest that formation of chlorophyll b in angiosperms can be different from that in green algae, but mimic it under a certain condition.

ACKNOWLEDGEMENTS

We thank Mr K. Furihata of this Institute for the ¹³C-NMR spectra. This work was supported by a grant-in-aid for special project research no. 56104008, from the Ministry of Education, Science and Culture, Japan.

REFERENCES

- [1] Beale, S.I. (1978) Annu. Rev. Plant Physiol. 29, 95-120.
- [2] Oh-hama, T., Seto, H., Otake, N. and Miyachi, S. (1982) Biochem. Biophys. Res. Commun. 105, 647-652.
- [3] Battersby, A.R. and McDonald, E. (1975) in: Porphyrins and Metalloporphyrins, pp. 61-122, Elsevier Scientific, Amsterdam, New York.
- [4] Oh-hama, T. and Hase, E. (1981) Plant Cell Physiol. 22, 747-757.
- [5] Ogawa, T. and Shibata, K. (1965) Photochem. Photobiol. 4, 193-200.
- [6] Omata, T. and Murata, N. (1980) Photochem. Photobiol. 31, 183-185.
- [7] Lötjönen, S. and Hynninen, P.V. (1981) Org. Magn. Reson. 16, 304-308.
- [8] Porra, R.J., Klein, O. and Wright, P.E. (1982) Biochem. Internatl. 5, 345-350.
- [9] Katz, J.J., Dougherty, R.C. and Boucher, L.J. (1966) in: The Chlorophylls (Vernon, L.P. and Seely, G.L. eds) pp. 185-251, Academic Press, London, New York.
- [10] Strouse, C.E., Kollman, V.H. and Matwiyoff, N.A. (1982) Biochem. Biophys. Res. Commun. 46, 328-334.
- [11] Shlyk, A.A. (1971) Annu. Rev. Plant Physiol. 22, 169-183.
- [12] Thornber, J.P. (1975) Annu. Rev. Plant Physiol. 26, 127-158.
- [13] Argyroudi-Akoyunoglou, J.H. and Castorinis, A. (1980) Arch. Biochem. Biophys. 200, 326-335.
- [14] Tanaka, A. and Tsuji, H. (1982) Biochim. Biophys. Acta 680, 265-270.